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## Citation for Chemical Breakthroughs



That chlorofluoromethanes photodissociate in the stratosphere,  
leading to the destruction of atmospheric ozone

Mario J. Molina and F. S. Rowland, *Nature* 1974, 249, 810-812.

### Stratospheric sink for chlorofluoromethanes : chlorine atom-catalysed destruction of ozone

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*Chlorofluoromethanes are being added to the environment in steadily increasing amounts. These compounds are chemically inert and may remain in the atmosphere for 40–150 years, and concentrations can be expected to reach 10 to 30 times present levels. Photodissociation of the chlorofluoromethanes in the stratosphere produces significant amounts of chlorine atoms, and leads to the destruction of atmospheric ozone.*

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It seems quite clear that the atmosphere has only a finite capacity for absorbing Cl atoms produced in the stratosphere, and that important consequences may result. This capacity is probably not sufficient in steady state even for the present rate of introduction of chlorofluoromethanes. More accurate estimates of this absorptive capacity need to be made in the immediate future in order to ascertain the levels of possible onset of environmental problems.

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